Oral Intake of Phosphorus Can Determine the Serum Concentration of 1,25-Dihydroxyvitamin D by Determining Its Production Rate in Humans

Anthony A. Portale, Bernard P. Hallowan, Margaret M. Murphy, and R. Curtis Morris, Jr.
General Clinical Research Center, Departments of Pediatrics and Medicine, University of California and Veterans Administration Medical Center, San Francisco, California 94143

Abstract

Changes in the oral intake of phosphorus could induce the reported changes in the serum concentration of 1,25-dihydroxyvitamin D (1,25-(OH)_2D) by inducing changes in its production rate (PR) or metabolic clearance rate (MCR), or both. To investigate these possibilities, we employed the constant infusion equilibrium technique to measure the PR and MCR of 1,25-(OH)_2D in six healthy men in whom the oral intake of phosphorus was initially maintained at 1,500 mg/70 kg body weight per d for 9 d, then restricted to 500 mg/d (coupled with oral administration of aluminum hydroxide) for 10 d, and then supplemented to 3,000 mg/d for 10 d. With phosphorus restriction, the serum concentration of 1,25-(OH)_2D increased by 80% from a mean of 38±3 to 68±6 pg/ml, P < 0.001; the PR increased from 1.8±0.2 to 3.8±0.6 μg/d, P < 0.005; the MCR did not change significantly. The fasting serum concentration of phosphorus decreased from 3.5±0.2 to 2.6±0.2 mg/dl, P < 0.01. With phosphorus supplementation, the serum concentration of 1,25-(OH)_2D decreased abruptly, reaching a nadir within 2 to 4 d; after 10 d of supplementation, the mean concentration of 27±4 pg/ml was lower by 29%, P < 0.01, than the value measured when phosphorus intake was normal. The PR decreased to 1.3±0.2 μg/d, P < 0.05; the MCR did not change significantly. The fasting serum concentration of phosphorus increased significantly, but only initially. These data demonstrate that in healthy men, reductions and increases in the oral intake of phosphorus can induce rapidly occurring, large, inverse, and persisting changes in the serum concentration of 1,25-(OH)_2D. Changes in the PR of 1,25-(OH)_2D account entirely for the phosphorus-induced changes in serum concentration of this hormone.

Introduction

1,25 dihydroxyvitamin D₃ (1,25-(OH)_2D₃) is the metabolite of vitamin D currently considered to be the most biologically active with respect to enhancement of bone resorption and intestinal absorption of calcium and phosphorus (1–4). The renal synthesis of 1,25-(OH)_2D₃ from its endogenous precursor, 25-hydroxyvitamin D₃, is catalyzed by 25-hydroxyvitamin D₃-1α-hydrox-

This work was presented in part at the National Meeting of the American Federation for Clinical Research, 1984.

Address reprint requests to Dr. Portale, 1202 Moffitt Hospital, University of California at San Francisco.

Received for publication 25 October 1984 and in revised form 1 August 1985.

J. Clin. Invest. © The American Society for Clinical Investigation, Inc. 0021-9738/86/01/0007/06 $1.00 Volume 77, January 1986, 7–12

Phosphorus Intake Determines the Production Rate of 1,25-Dihydroxyvitamin D

Methods

We determined the effect of restriction, then supplementation, of the oral intake of phosphorus on the serum concentration, PR, and MCR of 1,25-(OH)_2D in studies carried out in six healthy men (ages 26–40 yr). All studies were performed on the General Clinical Research Center under a protocol approved for use by the Committee on Human Research, University of California at San Francisco. Informed consent was obtained from each subject.

1. Abbreviations used in this paper: HPLC, high pressure liquid chromatography; 1α-hydroxylase, 25-hydroxyvitamin D₃-1α-hydroxylase; iPTH, immunoreactive parathyroid hormone; MCR, metabolic clearance rate; M-idPTH, midregion-iPTH; PR, production rate; PTH, parathyroid hormone.
Each subject received a constant whole food diet that provided, by calculation, 500 mg of phosphorus, 200 mg of calcium, 100 mg of magnesium, and 70 mg of sodium/70 kg body weight per d for 30 d. (All intakes are subsequently expressed per 70 kg body weight.) The intake of calcium and magnesium were maintained constant throughout by supplementing the whole food diet with orally administered calcium carbonate and magnesium sulfate (Table I). The intake of phosphorus was changed by the amount of phosphorus administered orally as a solution of neutral sodium and potassium phosphate (4:1 mixture of Na2HPO4/K2HPO4 and NaH2PO4/KH2PO4, 31 mg phosphorus, 0.9 mg sodium, and 0.9 mg potassium/5 ml) (Table I). During the first 9 d (5 precontrol, 4 d control), 1,000 mg/d of supplemental phosphorus was administered in divided doses with each meal. During the next 10 d, phosphorus was restricted by replacing the sodium and potassium phosphate supplement with an equivmolar amount of sodium and potassium chloride (0.9 meq sodium and 0.9 meq potassium/5 ml). Throughout this period only, aluminum hydroxide, 12 g/70 kg body weight per d, was administered in divided doses with each meal to decrease gut absorption of phosphorus. Throughout the subsequent final 10 d of study, dietary phosphorus was supplemented, 2,500 mg/d. This supplement provided 44 meq/d more sodium (and potassium) than provided by the diet and phosphate supplement during the first 19 d of the study. To keep the intakes of sodium (and potassium) constant throughout the entire study, additional sodium and potassium chloride, 44 meq/d each, was administered during the first 19 d (Table I). The basic diet provided, by calculation, 2,600 kcal/d, 9% as protein, 34% as fat, and 57% as carbohydrate.

Arterialized venous blood was drawn without stasis each day before breakfast at 8:30 a.m. for measurement of plasma concentration of ionized calcium and serum concentrations of creatinine, total calcium, phosphorus, magnesium, and 1,25-(OH)2D. On the last 2 d of each dietary period, the serum concentrations of 25-hydroxyvitamin D (25-OHD) and immunoreactive parathyroid hormone (iPTH) were also measured. Spontaneously voided urine was collected daily in 24-h pools for measurement of concentrations of creatinine, calcium, phosphorus, and magnesium. On the last day of each of the three dietary periods, we employed the constant isotope infusion equilibrium technique to determine the MCR and PR of 1,25-(OH)2D.

### Constant isotope infusion protocol

The constant infusion equilibrium method of Tait et al. (32, 33) was employed. Approximately 2.5 μCi of chromatographically purified 25-(OH)2D2 [9,26,27(3H)-vitamin D2 (2H1,25-(OH)2D2) (158 Ci/mmol, Amersham Corp., Arlington Heights, IL)] in 0.5 ml absolute ethanol were solubilized in a 10-ml sterile aliquot of the patients own serum, the latter dispersed in 0.9% NaCl, and the solution infused intravenously at a constant rate of 0.193 ml/min (~4,000 dpm/min) (Harvard Apparatus Co., Inc., S. Natick, MA). To assess the possibility that isotope might stick to the infusion apparatus (glass syringe and plastic tubing) during the 20-h period of infusion, we measured counts of H in aliquots of solution taken at the time isotope was prepared for infusion, and from the syringe and end of the plastic tubing at completion of the infusion some 21 h later. We found no evidence of significant sticking of isotope to the infusion apparatus during the period of infusion. In preliminary studies, we determined that stable serum concentrations of 1H-1,25-(OH)2D2 were attained and maintained after 16 to 18 h of constant infusion of radiolabeled hormone. In subsequent studies, radiolabeled hormone was infused for 20 h; during the final 3-h equilibrium period, blood was drawn at 30-min intervals for measurement of serum concentrations of both titrated and endogenous 1,25-(OH)2D. The total amount of 1H-1,25-(OH)2D administered over 20 h was ~0.005 μg, or 0.3% of the estimated daily renal production of the hormone.

### Laboratory methods

**Measurement of 1H-1,25-(OH)2D in serum.** Approximately 100 ng of chromatographically purified 1,25-(OH)2D3, kindly provided by Hoffman-La Roche, Inc., Nutley, NJ, in 20 μl of ethanol were added to each 2.0-ml serum sample to determine percentage of recovery. The serum was extracted twice with perchloric acid (0.1%, v/v), and the extract applied to a silica Sep-pak (Waters Inc., Milford, MA) in a solvent system of 0.1% isopropanol and 99.9% hexane. The Sep-pak was washed sequentially, first with 10 ml of isopropanol/hexane (2.5:97.5) and then with 7.5 ml of isopropanol/dichloromethane (4:96). 1,25-di-hydroxvitamin D was then eluted with 10 ml of isopropanol dichloromethane (10:90). The fraction was dried under N2, resublimated and applied to a radial compression separation system (Waters Inc.) using an 8-mm Si radial compression column (particle size, 5 μ) in a solvent system of isopropanol/hexane (8:92) at a flow rate of 4.5 ml/min. The content of 1H-1,25-(OH)2D in the collected fraction was determined by direct scintillation counting. Recovery of 1,25-(OH)2D3 added to the serum was accomplished by comparison of the ultraviolet absorbance maximum at 254 nm of the sample to that of a series of crystalline standards of 1,25-(OH)2D3. Recovery of 1,25-(OH)2D3 added to serum was 65–70%. Intraassay coefficient of variation of 1H-1,25-(OH)2D in serum was 6.8% at a tritium concentration of 150 dpm/ml.

To investigate whether titrated metabolites of vitamin D other than 1H-1,25-(OH)2D might be detected by our analytical procedures, preliminary studies were performed in which the concentration of 1H-1,25-(OH)2D during equilibrium was measured using two different chromatographic procedures, straight-phase high pressure liquid chromatography (HPLC) as described, and a combination of reverse-phase followed by straight-phase HPLC. In one subject, a large volume of blood was obtained during infusion equilibrium and the serum aliquoted into 10 2.0-ml fractions. Approximately 100 ng of 1,25-(OH)2D3 were added to each 2.0-ml sample, the serum extracted twice with diethyl ether, and the chromatographically purified 25-(OH)2D2, and the 0.9% NaCl infused the serum and removed the diethyl ether. From the solutions containing 1,25-(OH)2D were chromatographed using the HPLC procedure described. The other five fractions were applied first to a reverse-phase HPLC system using a μBondapak C 18 column (Waters Inc.) in a solvent system of water/methanol (25:75) at a flow rate of 2.0 ml/min, and then rechromatographed using the straight-phase radial compression separation system (Waters Inc.), as described. The content of 1H-1,25-(OH)2D in the collected fractions from both procedures was determined

### Table 1. Oral Intake of Minerals During Each Period of Study

<table>
<thead>
<tr>
<th>Mineral</th>
<th>Oral intake of phosphorus</th>
<th>Normal</th>
<th>Low</th>
<th>High</th>
</tr>
</thead>
<tbody>
<tr>
<td>Phosphorus (mg/d)</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total*</td>
<td>1,500</td>
<td></td>
<td>&lt;500</td>
<td>3,000</td>
</tr>
<tr>
<td>Supplement (Na/KPO4)</td>
<td>1,000</td>
<td></td>
<td>0</td>
<td>2,500</td>
</tr>
<tr>
<td>Calcium (mg/d)</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total*</td>
<td>850</td>
<td></td>
<td>850</td>
<td></td>
</tr>
<tr>
<td>Supplement (CaCO3)</td>
<td>650</td>
<td></td>
<td>650</td>
<td></td>
</tr>
<tr>
<td>Magnesium (mg/d)</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total</td>
<td>350</td>
<td></td>
<td>350</td>
<td></td>
</tr>
<tr>
<td>Supplement (MgSO4)</td>
<td>250</td>
<td></td>
<td>250</td>
<td></td>
</tr>
<tr>
<td>Sodium (meq/d)</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total</td>
<td>140</td>
<td></td>
<td>140</td>
<td></td>
</tr>
<tr>
<td>Supplement (Na/KCl)</td>
<td>44</td>
<td></td>
<td>73</td>
<td>0</td>
</tr>
<tr>
<td>(Na/KPO4)</td>
<td>29</td>
<td></td>
<td>0</td>
<td>73</td>
</tr>
</tbody>
</table>

Na/KPO4 is neutral sodium and potassium phosphate; Na/KCl is sodium and potassium chloride (see Methods).

* Total intake is calculated as the sum of those minerals provided as oral supplements in the amounts indicated, and as foodstuffs containing phosphorus, calcium, magnesium, and sodium in the amounts of 500, 200, 100 mg/d, and 70 meq/d, respectively.

‡ During the period of phosphorus restriction, aluminum hydroxide, 12 g/d, was administered in divided doses with meals to decrease gut absorption of phosphorus.
by direct scintillation counting. The two chromatographic procedures gave near identical values for the serum concentration of $^3$H-1,25-(OH)$_2$D: reverse-phase followed by straight-phase, 147±2 dpm/ml, combined recovery 39±2% (recovery of reverse-phase HPLC alone not measured); straight-phase, 153±3 dpm/ml, recovery 65±3%. Accordingly, we find no evidence that the analytical procedures employed detected tritiated metabolites of vitamin D other than $^3$H-1,25-(OH)$_2$D.

Serum concentrations of 1,25-(OH)$_2$D were measured in duplicate using a competitive protein binding assay (34) employing intestinal cytosol from normal vitamin D replete chicks. Minimum detection limits are <5 pg/assay tube; overall recovery ranged from 60 to 70%. Inter- and intraassay coefficients of variation of 1,25-(OH)$_2$D in serum were 13.4 and 12.6%, respectively, at a serum concentration of 31 pg/ml.

Serum concentrations of 25-OHD were measured as previously described (22). Serum concentrations of iPTH were measured by radioimmunoassay using two antisera: GP-IM, which has high affinity for PTH (1–84) and the midregion of the hormone, PTH (44–68), but low affinity for PTH (1–34), referred to hereafter as midregion-iPTH (M-iPTH), and CH-12M, which has high affinity for PTH (1–84), at least a 30-fold lower affinity for PTH (1–34), and no affinity for PTH (44–68) or carboxy-terminal fragments, referred to hereafter as intact-iPTH (35). Serum and urinary concentrations of calcium and magnesium were measured by atomic absorption spectrophotometry, serum and urinary concentrations of phosphorus by a modification of the Fiske-Subbarow method (36), and urinary creatinine by auto-analyzer. Plasma ionized calcium was measured in duplicate using the Nova 8 ionized calcium analyzer (Nova Biomedical, New York, MA).

**Data analysis.** The MCR of endogenous 1,25-(OH)$_2$D$_3$, defined as the volume of blood cleared completely and irreversibly of this hormone in unit time (33), is assumed to be equal to that of intravenously administered $^3$H-1,25-(OH)$_2$D. At infusion equilibrium, the MCR is calculated according to the relationship (33): MCR (ml/min) = (rate of infusion of $^3$H-1,25-(OH)$_2$D (dpm/ml)/serum concentration of $^3$H-1,25-(OH)$_2$D (dpm/ml)). The value for serum concentration of $^3$H-1,25-(OH)$_2$D used to calculate the MCR for each subject during each period of study is the mean of four or five separate determinations of $^3$H-1,25-(OH)$_2$D in serum obtained during equilibrium. Equilibrium was confirmed in each study by demonstrating that the slope of serum concentration of $^3$H-1,25-(OH)$_2$D against time was not significantly different from zero. The coefficient of variation of serum $^3$H-1,25-(OH)$_2$D during equilibrium was 6.6±0.8%, n = 17.

The PR of 1,25-(OH)$_2$D is calculated according to the relationship: PR (µg/d) = MCR (ml/min) × serum concentration of endogenous 1,25-(OH)$_2$D (µg/ml) × 1.440 min/d. The value for serum concentration of endogenous 1,25-(OH)$_2$D used to calculate the PR for each subject during each period of study is the mean of three separate determinations of 1,25-(OH)$_2$D in serum obtained during equilibrium. All values for MCR and PR are expressed per 70 kg body weight.

Data are presented as group means±SEM. Statistical analysis was performed using repeated measurements analysis of variance; changes from control (normal dietary phosphorus) were analyzed using the paired t-test. Statistical significance was defined as P < 0.025, after the Bonferroni correction for two comparisons (37).

**Results**

**Serum 1,25-(OH)$_2$D.** When the oral intake of phosphorus was normal, 1,500 mg/d, the serum concentration of 1,25-(OH)$_2$D in the six subjects ranged from 29 to 48 pg/ml, with the mean being 38±3 pg/ml. When phosphorus was restricted, the serum concentration of 1,25-(OH)$_2$D increased immediately and progressively, and after 6 to 8 d reached a relatively constant value of 63–75 pg/ml (Fig. 1). After 10 d of restriction, the mean value, 68±6 pg/ml, was 80% greater (P < 0.001) than that obtained with a normal phosphorus intake. When phosphorus was supplemented, the serum concentration of 1,25-(OH)$_2$D decreased abruptly to a nadir of 13–18 pg/ml in 2–4 d. Over the subsequent 2–3 d, the value increased somewhat to attain a relatively constant value of 27±4 pg/ml; this value was 60% lower than that obtained with phosphorus restriction, and 29% lower (P < 0.01) than that obtaining with a normal phosphorus intake. The serum concentration of 25-OHD, 13±2 ng/ml, did not change significantly when phosphorus was either restricted, 12±3 ng/ml, or supplemented, 16±4 ng/ml.

**PR and MCR of 1,25-(OH)$_2$D.** To determine whether the phosphorus-induced changes in serum concentration of 1,25-(OH)$_2$D were due to changes in its PR or MCR, or both, we employed the infusion equilibrium technique to estimate these rates after each subject had been maintained on each of the three intakes of phosphorus for 10 d. When the intake of phosphorus was normal, the MCR ranged from 25 to 39 ml/min per 70 kg, with a mean of 34±2 ml/min. The PR ranged from 1.3 to 2.5 µg/d per 70 kg, with a mean of 1.8±0.2 µg/d (Fig. 2).

When phosphorus was restricted, the increase in serum concentration of 1,25-(OH)$_2$D was attended by an increase in the PR in each subject; the mean value increased to 3.8±0.6 µg/d (P < 0.005). The metabolic clearance rate did not change significantly. When phosphorus was supplemented, the PR of 1,25-(OH)$_2$D decreased in each subject, and for the group from 3.8±0.6 to 1.3±0.2 µg/d, a decrease of 60%; the decrease from the normal phosphorus intake was 27% (P < 0.05). The MCR did not change significantly.

**Serum phosphorus, calcium, magnesium, and iPTH.** When phosphorus was restricted, the fasting serum concentration of phosphorus decreased from a mean of 3.5±0.2 to 2.6±0.2 mg/dl (P < 0.01) (Table II, Fig. 1). When phosphorus was supplemented, the fasting serum concentration of phosphorus increased to a maximal value after 1 d of supplementation, but then decreased so that during the final 4 d of supplementation, the mean value of 3.6±0.2 mg/dl was not significantly different from that measured with a normal phosphorus intake.

The mean serum concentrations of total and ionized calcium did not differ significantly on each of the three intakes of phosphorus; however, during the transition from the low to the high phosphorus intake, the serum concentration of ionized calcium

![Figure 1. Effect of changes in the oral intake of phosphorus on the fasting serum concentrations of 1,25-(OH)$_2$D and phosphorus in six healthy men. The bracketed points depict mean values±SEM.](image-url)

**Phosphorus Intake Determines the Production Rate of 1,25-Dihydroxyvitamin D**
decreased transiently. The serum concentration of magnesium decreased slightly but significantly when phosphorus was restricted. The serum concentrations of neither M-iPTH nor intact-iPTH differed significantly on each of the three intakes of phosphorus.

Urinary phosphorus, calcium, and magnesium. Urinary excretion of phosphorus decreased and increased rapidly in response to the decrease and increase, respectively, in phosphorus intake (Table II); renal adaptation was essentially complete within 48 h of either maneuver. Urinary excretion of calcium increased immediately and progressively after phosphorus was restricted, and after 10 d had increased by about threefold. After 10 d of phosphorus supplementation, calcium excretion was slightly but significantly lower than the control value. The changes from control (normal phosphorus intake) in urinary excretion of calcium induced by restriction and supplementation of phosphorus varied directly and significantly with the changes from control induced in serum concentration of 1,25-(OH)2D, when the values in the six subjects were analyzed as a single set (r = 0.91, P < 0.001). Changes in urinary excretion of magnesium were similar in direction to, but of a lesser magnitude than, those of urinary calcium.

Discussion

The results of the present study confirm that in normal adult subjects (18–21), as in children with moderate renal insufficiency (22), restriction of the oral intake of phosphorus combined with ingestion of aluminum hydroxide induces an increase in the serum concentration of 1,25-(OH)2D. When phosphorus was restricted for 10 d, the serum concentration of 1,25-(OH)2D increased immediately and progressively, and within a week, reached a stable maximal value almost twice the value when phosphorus intake was normal (Fig. 1). The pattern of the increase in serum concentration of 1,25-(OH)2D induced by phosphorus restriction is like that previously observed in healthy adult subjects (19–21), although the magnitude of the increase we observed (8%) is greater than that reported to occur after either 4, 10, or 18 d of phosphorus restriction (29, 46, 32%, respectively) (19–21). It seems likely that the greater increase in serum 1,25-(OH)2D observed in the present study reflects a more severe restriction of dietary phosphorus (19, 21), and possibly also differences in the sex and age of study subjects (20).

The results of the present study demonstrate that in healthy men, supplementation of dietary phosphorus induces a decrease in the serum concentration of 1,25-(OH)2D. When phosphorus was supplemented for 10 d (after its restriction), the serum concentration of 1,25-(OH)2D decreased from a maximal value of 68 pg/ml to a nadir of ~15 pg/ml within 2 d; thereafter, the concentration increased somewhat to attain a near constant value 60% lower than that obtaining with phosphorus restriction, and ~30% lower than the value obtaining with normal phosphorus intake. These findings demonstrate that in man, persisting changes in the oral intake of phosphorus can induce rapidly occurring, large, inverse, and persisting changes in the serum concentration of 1,25-(OH)2D.

We employed the isotope infusion equilibrium technique to determine whether the phosphorus-induced changes in serum concentration of 1,25-(OH)2D reflected changes in its PR or MCR, or both. When phosphorus intake was normal, 1,500 mg/d, the MCR of 1,25-(OH)2D was estimated to be 34±2 ml/min in the six subjects. We have previously reported that in healthy men, the endogenous serum concentration of 1,25-(OH)2D varies by <20% during a 24-h period (38); in the present study the concentration of endogenous hormone was virtually constant during the 3-h infusion equilibrium period. Hence, the PR of 1,25-(OH)2D can be reliably estimated from measurements of MCR and serum concentration of endogenous hormone; the mean value for production rate was 1.8±0.2 pg/min. In each of the six subjects studied, the individual values for MCR and PR were clustered tightly about their respective mean values. These mean values are in good agreement with those reported in normal adult subjects, as determined by the equilibrium infusion technique (31), and are similar to (39) or somewhat higher than (40, 41) those determined by the bolus injection technique.

When phosphorus was restricted and then supplemented, the observed increase and decrease in serum concentration of 1,25-(OH)2D was associated with an increase and decrease, respectively, in the PR of the hormone, but not with a significant change in its MCR. Thus, these findings demonstrate that in healthy men, changes in the PR of 1,25-(OH)2D account entirely for the phosphorus-induced changes in serum concentration of this hormone. The present findings accord with studies in the intact chick and rat in which phosphorus restriction increased the activity of 1-hydroxylase (16) and the synthetic rate of 1,25-(OH)2D (25, 26), when measured in vitro in renal cortical tissue. The present studies are also consistent with previous studies in phosphorus-deprived healthy men in whom the rate of disappearance of radiolabeled 25-OH-D3 from plasma was increased,
Table II: Changes in Blood and Urine Composition Induced by Restriction and Supplementation of Oral Intake of Phosphorus in Six Healthy Men

| Phosphorus intake | Total Ca | Urea | Pi | Mg | Ca** | Ca/PI | El-P | Na | K | P | Phosphorus intake | Total Ca | Urea | Pi | Mg | Ca** | Ca/PI | El-P | Na | K | P |
|--------------------|---------|------|----|----|------|-------|------|-----|---|---|--------------------|---------|------|----|----|------|-------|------|-----|---|---|---|
| Normal             | 4.58±0.08 | 9.5±0.2 | 2.6±0.2 | 1.8±0.04 | 0.1±0.1 | 0.1±0.2 | 0.1±0.03 | 0.01 | NS | NS | Normal             | 4.58±0.08 | 9.5±0.2 | 2.6±0.2 | 1.8±0.04 | 0.1±0.1 | 0.1±0.2 | 0.1±0.03 | 0.01 | NS | NS |
| Restricted         | 4.58±0.01 | 9.4±0.2 | 2.6±0.2 | 1.8±0.04 | 0.1±0.1 | 0.1±0.2 | 0.1±0.03 | 0.01 | NS | NS | Restricted         | 4.58±0.01 | 9.4±0.2 | 2.6±0.2 | 1.8±0.04 | 0.1±0.1 | 0.1±0.2 | 0.1±0.03 | 0.01 | NS | NS |
| From normal        | 0       | 0     | 0   | 0   | 0    | 0     | 0    | 0   | 0 | 0 | From normal        | 0       | 0     | 0   | 0   | 0    | 0     | 0    | 0   | 0 | 0 |
| P                  | NS      | NS    | NS  | NS  | NS   | NS    | NS   | NS  | NS | NS | P                  | NS      | NS    | NS  | NS  | NS   | NS    | NS   | NS  | NS | NS |
| Supplemented       | 4.3±0.06 | 9.7±0.2 | 2.6±0.2 | 1.8±0.04 | 0.1±0.1 | 0.1±0.2 | 0.1±0.03 | 0.01 | NS | NS | Supplemented       | 4.3±0.06 | 9.7±0.2 | 2.6±0.2 | 1.8±0.04 | 0.1±0.1 | 0.1±0.2 | 0.1±0.03 | 0.01 | NS | NS |
| From normal        | 0       | 0     | 0   | 0   | 0    | 0     | 0    | 0   | 0 | 0 | From normal        | 0       | 0     | 0   | 0   | 0    | 0     | 0    | 0   | 0 | 0 |
| P                  | NS      | NS    | NS  | NS  | NS   | NS    | NS   | NS  | NS | NS | P                  | NS      | NS    | NS  | NS  | NS   | NS    | NS   | NS  | NS | NS |

Values are mean±SEM; n, six subjects. Values used in calculation of the depicted means are the average of four separate determinations made on each of the last 4 d of each dietary period, unless otherwise indicated. Cor, creatinine clearance. * Two determinations during each dietary period.

which suggests that the rate of its conversion to 1,25-(OH)_{2}D was increased (42). The present studies accord with the proposal that in children with moderate renal insufficiency, changes in the renal production of 1,25-(OH)_{2}D underlie the reduction in its plasma concentration observed when dietary phosphorus is normal or supplemented (22, 43), and its increase to normal values observed when phosphorus is restricted (22).

The present studies do not address the question of the mechanism by which changes in the oral intake of phosphorus induce changes in the PR of 1,25-(OH)_{2}D. We observed that phosphorus restriction induced a modest but sustained decrease in the fasting serum concentration of phosphorus. Yet, Maierhofer et al. (21) reported that phosphorus restriction in healthy men induced a 30% increase in serum concentration of 1,25-(OH)_{2}D despite the absence of a sustained decrease in fasting serum concentration of phosphorus or detectably more negative phosphorus balance (21). Conversely, in the present study, during the transition from the low to high phosphorus intake when the serum level of 1,25-(OH)_{2}D decreased abruptly by ~80%, the fasting serum concentration of phosphorus nearly doubled. But fasting hyperphosphatemia was not sustained, and after 10 d of supplementation the fasting serum level of phosphorus was not higher than the control value, whereas the serum concentration and PR of 1,25-(OH)_{2}D had decreased by ~30%. Changes in phosphorus intake may induce changes in the serum concentration of phosphorus later in the day, or may induce changes in intracellular concentration or transepithelial processing of phosphorus that affect the production rate of 1,25-(OH)_{2}D even in the absence of changes in the serum concentration of phosphorus.

Acknowledgments

We thank the nursing staff of the General Clinical Research Center. We also thank Margaret Castro, Vivian Ho, and Bernadine Serena for expert technical assistance, and Andrea Marcellano for help in preparation of this manuscript.

This work was supported by grants from the National Institutes of Health (National Institute of Arthritis, Metabolism, and Digestive Diseases, AM 21354 and AM 30513), the Division of Research Resources (General Clinical Research Center, RR 00079), from the Veterans Administration, and by generous gifts from the Church and Dwight Corp. and the Emil Mosbacher, Jr. Foundation.

References


20. Lefkowitz, E. G., R. Kumar, and H. Heath III. 1983. Hyperphos-


22. Portale, A. A., B. E. Booth, B. P. Halloran, and R. C. Morris, Jr. 1984. Effect of dietary phosphorus on circulating concentrations of 1,25-


36. Fiske, C. H., and Y. Subbarow. 1925. The colorimetric deter-


39. Seeman, E., R. Kumar, G. G. Hunder, M. Scott, H. Heath III, and B. L. Rigs. 1980. Production, degradation, and circulating levels of 1,25-dihydroxyvitamin D in health and in chronic glucocorticoid exces-


A. A. Portale, B. P. Halloran, M. M. Murphy, and R. C. Morris, Jr.