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Pathophysiological Concentrations of Glucose Promote Oxidative Modification of Low Density Lipoprotein by a Superoxide-dependent Pathway

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Abstract

Oxidized lipoproteins may be important in the pathogenesis of atherosclerosis. Because diabetic subjects are particularly prone to vascular disease, and glucose autoxidation and protein glycation generate reactive oxygen species, we explored the role of glucose in lipoprotein oxidation. Glucose enhanced low density lipoprotein (LDL) oxidation at concentrations seen in the diabetic state. Conjugated dienes, thiobarbituric acid reactive substances, electrophoretic mobility, and degradation by macrophages were increased when LDL was modified in the presence of glucose. In contrast, free lysine groups and fibroblast degradation were reduced. Although loss of reactive lysine groups could be due to either oxidative modification or nonenzymatic glycation of apolipoprotein B-100, inhibition of lipid peroxidation by the metal chelator, diethylenetriamine pentaacetic acid, blocked the changes in free lysines. Thus, glycation of lysine residues is unlikely to account for the alterations in macrophage and fibroblast uptake of LDL modified in the presence of glucose. Glucose-mediated enhancement of LDL oxidation was partially blocked by superoxide dismutase and nearly completely inhibited by butylated hydroxytoluene. These findings indicate that glucose enhances LDL lipid peroxidation by an oxidative pathway involving superoxide and raise the possibility that the chronic hyperglycemia of diabetes accelerates lipoprotein oxidation, thereby promoting diabetic vascular disease. (J. Clin. Invest. 1994. 94:771–778.) Key words: peroxidation • superoxide • hydrogen peroxide • macrophages • diabetes

Introduction

Atherosclerosis is a leading cause of morbidity and mortality in patients suffering from diabetes mellitus. Although other disorders associated with vascular disease, such as hypertension, central obesity, and hyperlipidemia, frequently coexist with the diabetic state (1), they appear inadequate to explain the increased risk of atherosclerosis. These observations suggest that additional factors may be involved in the accelerated macrovascular disease of diabetes.

Epidemiological, clinical, and genetic investigations indicate that elevated levels of low density lipoprotein (LDL) are another important risk factor for coronary artery disease. Although LDL levels are frequently normal in diabetic patients, hyperglycemic or other metabolic alterations secondary to the diabetic state may alter lipoproteins to a form that promotes atherogenesis. One such modification may be the nonenzymatic glycation of LDL which has been shown in some (2), but not all, studies (3) to be taken up rapidly by macrophages in culture. The unregulated uptake of modified lipoproteins by macrophages is likely to play a pivotal role in cholesterol accumulation, a key early event in lesion formation. Autoantibodies against glycated LDL (4) also might result in the formation of immune complexes, which have been postulated to stimulate macrophage foam cell formation (5). Recently, there has been considerable interest in oxidized LDL, another modification of lipoproteins that may play an important role in atherogenesis (6, 7). LDL can be oxidatively damaged in vitro by endothelial cells, monocyte-macrophages, and arterial smooth muscle cells (for review see reference 7). Lipoxygenases may be involved in the oxidative modification of LDL by endothelial cells and macrophages (7, 8), whereas superoxide plays an important role in LDL oxidation by arterial smooth muscle cells and activated monocytes (9, 10). Oxidized LDL is thought to promote atherogenesis by a multitude of different mechanisms (7, 11).

Indirect evidence suggests that lipoprotein oxidation may be increased in diabetes. First, increased levels of thiobarbituric acid reactive substances (TBARS)1 have been demonstrated in plasma from diabetic subjects (12). Although TBARS increase during oxidative modification of lipoproteins, their presence in plasma is not specific for products of lipid peroxidation. Second, plasma lipoproteins isolated from diabetic rats have increased TBARS and are cytotoxic to cultured cells (13), which suggests that they may have been oxidatively modified in vivo. Both lipoprotein TBARS content and cytotoxicity were reduced either by antioxidants or by treatment of hyperglycemia with insulin (13). Finally, glucose autoxidation (14, 15) and nonenzymatic protein glycation (15, 16) can result in the generation of oxygen-free radicals, such as superoxide, which stimulate lipoprotein oxidation (9, 10). Indeed, a recent study has suggested that LDL lipid peroxidation is enhanced by high concentrations of glucose (17).

In the current studies we have explored the relationships between elevated levels of glucose, the hallmark of the diabetic state, and the oxidative modification of LDL. We show that glucose promotes lipoprotein oxidation by a pathway that involves superoxide. Human monocyte-derived macrophages rap-

1. Abbreviations used in this paper: BHT, butylated hydroxytoluene; DTPA, diethylenetriamine pentaacetic acid; TBARS, thiobarbituric acid reactive substances; TNBS, trinitrobenzenesulfonic acid.
idly take up and degrade LDL modified in the presence of glucose, suggesting a potential mechanism whereby hyperglycemia favors foam cell formation and atherogenesis.

**Methods**

**Materials.** $^{125}$I-Na was obtained from DuPont/New England Nuclear (Boston, MA). Ficoll-paque was purchased from Pharmacia Fine Chemicals (Piscataway, NJ). Butylated hydroxytoluene (BHT), diethylentriamine pentaaetic acid (DTPA), SOD, catalase, and glucose were obtained from Sigma Chemical Company (St. Louis, MO). Trinitrobenzene sulfonic acid (TNBS) was obtained from Pierce Chemical Company (Rockford, IL). DMEM and RPMI 1640 culture medium, fetal calf serum, penicillin-streptomycin, and l-glutamine were from Gibco BRL (Gaithersburg, MD).

**Lipoproteins.** LDL was prepared from human plasma (EDTA 4 mM) that was treated immediately after separation from red cells with 1 µM (final concentration) phenylmethylsulphonyl fluoride. The density of the plasma was increased to 1.30 g/ml by the addition of solid KBr, and LDL (density = 1.019–1.063 g/ml) was isolated by discontinuous density gradient ultracentrifugation in a vertical rotor (VT-50; Beckman Instruments, Inc., Fullerton, CA) by the method of Chung et al. (18). All solutions used for lipoprotein preparation contained 1 mM EDTA. After one wash at $d = 1.063$ g/ml, the isolated LDL was extensively dialyzed against phosphate-buffered solution under $N_2$ at 4°C against 150 mM NaCl and 1 mM EDTA (pH 7.4) for 24 h, stored at 4°C under $N_2$ in the dark, and used within 2 wk.

**LDL modification.** After the removal of the EDTA by dialysis against phosphate-buffered solution under $N_2$ at 4°C, LDL (200 µg of protein/ml) was incubated with the indicated additions in 100 mM sodium phosphate buffer (pH 7.4) containing 1 µM EDTA at 37°C under 5% CO$_2$/95% air. Oxidation was arrested by refrigeration and by the addition of 50 µM DTPA and 25 µM BHT (9). The extent of LDL lipid peroxidation was evaluated using diene conjugation, iodide reactive material, and by the appearance of TBARS (19). Conjugated dienes were assessed by monitoring the increase in absorbance at 232 nm as described by Esterbauer et al. (20) using 160 µg of LDL protein in a final volume of 0.8 ml. Lipid peroxides were measured directly by means of the iodometric assay described by El-Saadani et al. (21) using 40 µg of LDL protein in a final volume of 1 ml. TBARS were measured as described by Buege and Aust (22) using 40 µg of LDL protein in a final volume of 1.5 ml. Preliminary experiments indicated that neither glucose (at concentrations of up to 25 mM) nor copper (at micromolar concentrations) interfered with the assays (data not shown). The coefficients of variation for the conjugate diene, lipid peroxides, and TBARS assays were 2.6, 4.2, and 10.9%, respectively. LDL was acetylated as described by Basu et al. (23) and radiolabeled by the iodine monochloride method as modified for lipoproteins by Bilheimer et al. (24) using $^{125}$I-Na.

**Cells.** Human monocyte-derived macrophages were isolated by density gradient centrifugation by the method of Boyum (25). The mononuclear cell band was washed twice at 4°C in RPMI 1640 medium and then plated at $3 \times 10^5$ cells/16-mm well at 37°C in 5% CO$_2$/95% air in the same medium containing 20% (vol/vol) autologous serum, 100 U/ml penicillin, 100 µg/ml streptomycin, and 2 mM glucose (26). Monocyte-derived macrophages were fed twice weekly and used within 7–10 d of plating. Human skin fibroblasts were cultured from punch biopsies from the medial thigh of normal volunteers as described previously (26). Subcultures were used between passages 4 and 12. The cells were plated at $7 \times 10^4$ cells/35-mm dish at 37°C in 5% CO$_2$/95% air in DME supplemented with 10% calf serum, 100 U/ml penicillin, and 100 µg/ml streptomycin. After 5 d in culture, the medium of the cells was changed to DME supplemented with 10% lipoprotein-deficient human serum (prepared by ultracentrifugation at $d = 1.25$ g/ml) and used after 48 h of incubation.

**Metabolism of lipoproteins by cells.** Before incubation with fibroblasts and macrophages, $^{125}$I-labeled lipoproteins were dialyzed against 150 mM NaCl with 50 µM DTPA (pH 7.4) to remove the reaction mixture used for modification. Uptake and degradation of the $^{125}$I-labeled lipoprotein (150–250 cpm/ng protein) were assessed after a 5-h incubation with cells at 37°C. LDL degradation products in the incubation medium were assayed by measurement of trichloroacetic acid–soluble

![Figure 1.](https://example.com/figure1.png)
oxidative modification, experiments were performed in which lipid peroxidation was blocked by the transition metal chelator, DTPA. The extent of lipoprotein modification was determined by measurement of TBARS, which measures soluble bifunctional aldehydes derived from lipid peroxidation, and of TNBS reactivity, which measures free amine groups on protein-associated lysyl residues. In the presence of both glucose and copper, but in the absence of DTPA, TBARS progressively increased (Fig. 4a) while TNBS reactivity progressively decreased (Fig. 4b). Similar results were observed when LDL was incubated with either glucose or copper alone, or together (Fig. 4). These findings could reflect derivatization of protein amino groups by either lipid peroxidation products or by nonenzymatic glycation. When the same experiment was performed in the presence of DTPA, which inhibits lipid peroxidation, no changes were observed in either TBARS (Fig. 4c) or TNBS reactivity (Fig. 4d). Because glycation is largely independent of transition metal ions, these results suggest that modification of apolipoprotein B-100 was due to oxidation. In the absence of DTPA, the TBARS content of LDL demonstrated an inverse linear relationship with TNBS reactivity (Fig. 5), providing further evidence that protein amino groups were reacting with lipid oxidation products.

To determine the mechanism(s) by which glucose facilitates the oxidative modification of LDL, we studied the effect of several scavengers of reactive oxygen species. Stimulation of lipid peroxidation by glucose in the presence of added copper (0.5 μM) was partially inhibited either by SOL (30 μg/ml), which catalyzes the breakdown of superoxide, or by catalase (50 μg/ml), which catalyzes the breakdown of hydrogen peroxide (Fig. 6d). The effect of these enzymes on the inhibition of LDL oxidation was additive. Heat inactivation of either SOL or catalase eliminated their inhibitory activity (data not shown). BHT markedly inhibited glucose-mediated LDL oxidation. SOL and catalase together almost completely inhibited LDL oxidation in the absence of either glucose (Fig. 6b) or copper alone (Fig. 6c). Taken together, these results indicate that the glucose- and copper-induced lipid peroxidation of LDL is likely to be dependent on a reaction pathway that requires superoxide and hydrogen peroxide.

To explore the possible role of superoxide and hydrogen peroxide in glucose-stimulated oxidation, similar studies were performed with LDL incubated for 120 h with 15 mM glucose in the absence of added copper. As with glucose and copper together, SOL partially prevented LDL oxidation, measured as the increase in conjugated dienes, lipid peroxides, and TBARS (Table I). In contrast, catalase alone had little inhibitory effect. Heat-inactivated SOL failed to block glucose-induced LDL oxidation, suggesting that enzymatic activity rather than nonspecific effects, such as metal ion binding, accounted for inhibition
of lipid peroxidation (31, 32). These results indicate that LDL oxidation by glucose, when excess copper is not added, is likely to involve superoxide but not hydrogen peroxide.

We next investigated the cellular metabolism of the modified lipoproteins using cultured human monocyte-derived macrophages and skin fibroblasts. The uptake and degradation of modified 125I-LDL by macrophages increased in parallel with the increase in lipid peroxidation (Fig. 7 a). In contrast, the rates of degradation of these lipoproteins by fibroblasts occurred in the reverse order: LDL was degraded most rapidly, acetylated LDL was degraded least rapidly, while the oxidatively modified forms of LDL were degraded at intermediate rates (Fig. 7 b). The free protein amino content of the lipoproteins showed an inverse correlation with the rate of degradation of these lipoproteins by macrophages (Fig. 8 a) and a positive correlation with the rate of degradation by skin fibroblasts (Fig. 8 b). Human macrophages exhibit numerous scavenger receptors but are relatively poor in LDL-receptor activity, while fibroblasts are rich in LDL receptors but lack scavenger receptor activity (33). This suggests that modification of LDL by glucose and copper results in increased recognition and uptake by the scavenger receptor and decreased recognition and uptake by the LDL receptor.

**Discussion**

These studies demonstrate that physiologically relevant concentrations of glucose enhance the oxidation of LDL in a cell-free system. Although the addition of glucose alone led to increased LDL oxidation, this effect was magnified in the presence of low concentrations of copper, a transition metal that is redox-active and catalyzes lipid peroxidation (34, 35). The effect of glucose on LDL oxidation was most pronounced at glucose concentrations that span the range seen in the diabetic state.

To determine the mechanism(s) whereby glucose stimulated LDL lipid peroxidation, several inhibitors of reactive oxygen species were tested. LDL oxidation by glucose in the absence of added metal ions was inhibited by SOD but not by catalase. The exact mechanism for the acceleration of LDL lipid peroxidation by glucose is unknown but is likely to involve superoxide
metal ions may be endogenously bound to LDL (36) and are ubiquitously present in the buffers used for biochemical studies (34). It is noteworthy that low levels of protein-bound metal ions have been implicated in the oxidative damage of tissue during aging (37) and that products generated by metal-catalyzed glycoxidation reactions accumulate in the proteins of diabetic patients (15). Thus, protein-bound metal ions may catalyze oxidation reactions in vivo.

The pathway for LDL oxidation was apparently different when both glucose and copper were present in the reaction mixture. Under these conditions, LDL oxidation was partially inhibited by either SOD or catalase, and the enzymes together completely blocked lipid peroxidation. Thus, both superoxide and hydrogen peroxide appear to play a role in the stimulation of LDL oxidation by glucose together with copper. One potential mechanism consistent with these observations is generation of hydroxyl radicals (HO') by superoxide-driven Haber-Weiss chemistry (Eq. 1-3) (35):

\[ \text{O}_2^- + M^+ \rightarrow \text{O}_2 + M^{n-1} \]  
\[ \text{M}^{n-1} + \text{H}_2\text{O}_2 \rightarrow \text{M}^n + \text{HO}^- + \text{OH}^- \]  
\[ \text{Sum } \text{O}_2^- + \text{H}_2\text{O}_2 \rightarrow \text{O}_2 + \text{HO}^- + \text{OH}^- \]

Consistent with this hypothesis, hydroxyl radicals are a potent initiator of lipid peroxidation and are known to damage both proteins and lipids by H' abstraction. However, glucose reacts rapidly with hydroxyl radical (rate constant \(1 \times 10^9 \text{ M}^{-1} \cdot \text{s}^{-1}\)) (38), suggesting that initiation of lipid peroxidation by this pathway would be site specific. The recent demonstration that LDL binds copper ions (36) raises the interesting possibility that such oxidative reactions might take place at the interface between the aqueous and lipid phases of the lipoprotein. Alternatively, Aust and Minotti (39) have proposed that an optimum mixture of reduced and oxidized iron is necessary for initiation of liposomal peroxidation. According to this theory, \(\text{O}_2^-\) and

**Figure 5.** Relationship between oxidative modification and free amino groups on lipoproteins. LDL was incubated for 48 h under the conditions described in the legend to Fig. 4 in the absence of DTPA. The extent of lipid peroxidation was assessed as TBARS and free amino groups as TNBS reactivity as described in Methods. Values shown are means of duplicate determinations. The experiment was repeated three times. LDL no additions; LDL-G, 25 mM glucose; LDL-Cu, 0.5 \(\mu\)M CuSO4; LDL-G-Cu, 25 \(\mu\)M glucose, 0.5 \(\mu\)M CuSO4.

**Figure 6.** Effect of inhibitors of oxidation on glucose- and copper-mediated modification of LDL. LDL was incubated with no addition (Fig. 6 a), with glucose (25 mM) alone (Fig. 6 b), CuSO4 (0.5 \(\mu\)M) alone (Fig. 6 c), and glucose and CuSO4, (Fig. 6 d) for the indicated times in the absence of inhibitor (○) or in the presence of 30 \(\mu\)g/ml SOD (●), 50 \(\mu\)g/ml catalase (▲), SOD and catalase (▼), or 25 \(\mu\)M BHT (□), after which the lipid peroxide content was determined by the iodometric reaction as described in Methods. Values shown are means of duplicate determinations. The results shown are representative of three similar experiments.
The complete system consisted of LDL (200 μg/ml), 15 mM glucose, and 1 μM EDTA in sodium phosphate buffer (100 mM, pH 7.4). SOD and catalase were heat-inactivated by autoclaving and boiling for 10 min, respectively. After incubation for 120 h at 37°C in 95% air/5% CO₂, the reaction mixture was assayed for lipid peroxides and TBARS as described in Methods. Values represent the mean±SD of quadruplicate determinations. The results are representative of three similar experiments. *A₃₅₅ × 1,000; †A₃₃₂ × 1,000; ‡A₂₃₄ × 1,000 after a 1:5 dilution of the sample.

Derivatization of lysine ε-amino residue on apolipoprotein B-100 alters the receptor recognition of LDL and thereby controls its uptake and degradation by cells (33, 43). When lysine residues are modified by acetylation or by reaction with malondialdehyde, LDL becomes a ligand for the scavenger receptor on macrophages but loses recognition by the LDL receptor on human fibroblasts (33, 43). Derivatization of lysine amines by reactive aldehydes generated during lipid peroxidation may likewise be important in mediating the binding of oxidatively modified LDL to the scavenger receptor (44). Similar reactions might cause LDL modified in the presence of glucose to gain scavenger receptor recognition. Alternatively, nonenzymatic protein glycation may alter cell-surface recognition of the modified lipoprotein (2–4). To investigate the relative contributions of these pathways to alterations in apolipoprotein B-100 of modified LDL in the presence of glucose, we monitored the reactivity of protein amino groups with TNBS. This reagent predominantly reacts with free amines on lysines that are accessible to the aqueous phase (28). When LDL was incubated with glucose and copper, over 50% of TNBS reactivity was lost. The decrease in reactive amino groups was associated with an increase in lipid peroxidation products as measured by TBARS. Inclusion of DTPA, which chelates the metal ions necessary for propagation of lipid peroxidation in a redox inactive form (9), blocked both loss of TNBS reactivity and formation of TBARS. Although the slow process of nonenzymatic protein glycation can be inhibited slightly by metal ion chelation (45), we observed complete inhibition both of loss of TNBS reactivity and of increase in lipid peroxidation in the presence of DTPA. This observation strongly suggests that loss of free amino groups is likely to reflect oxidative damage to apolipoprotein B-100 by a glucose-dependent mechanism. However, we cannot exclude the possibility that nonenzymatic glycation of LDL, below the level of detection of the TNBS assay, contributed to the oxidative modification by generating reactive oxygen species.

The cellular metabolism of glucose-modified LDL was examined using cultured human skin fibroblasts and human monocyte-derived macrophages. Uptake of the modified LDL by fibroblasts decreases with increasing oxidative modification. The uptake and degradation of LDL oxidized in the presence of both glucose and copper by fibroblasts were considerably inhibited relative to unmodified LDL, consistent with decreased recognition of modified LDL by the scavenger receptor.

**Table I. Effect of Scavengers of Reactive Oxygen Species on Glucose-stimulated LDL Oxidation**

<table>
<thead>
<tr>
<th>Lipid peroxides*</th>
<th>TBARS† (percentage of control)</th>
<th>Conjugated dienes‡</th>
</tr>
</thead>
<tbody>
<tr>
<td>Complete system</td>
<td>168±3 (100)</td>
<td>871±7 (100)</td>
</tr>
<tr>
<td>Complete system minus glucose</td>
<td>76±2 (45)</td>
<td>553±2 (63)</td>
</tr>
<tr>
<td>Complete system plus SOD (5 μg/ml)</td>
<td>34±2 (20)</td>
<td>389±2 (45)</td>
</tr>
<tr>
<td>Heat-inactivated SOD</td>
<td>172±7 (102)</td>
<td>843±11 (97)</td>
</tr>
<tr>
<td>Catalase (5 μg/ml)</td>
<td>150±2 (89)</td>
<td>826±5 (95)</td>
</tr>
<tr>
<td>Heat-inactivated catalase</td>
<td>184±3 (109)</td>
<td>864±7 (99)</td>
</tr>
<tr>
<td>SOD plus catalase</td>
<td>49±3 (29)</td>
<td>484±10 (56)</td>
</tr>
</tbody>
</table>

H₂O₂ modulate lipid peroxidation by altering the amounts of Fe²⁺ and Fe³⁺: O₂⁻ would reduce Fe³⁺ to Fe²⁺ while H₂O₂ would oxidize Fe²⁺ to Fe³⁺. At an optimal ratio of reduced to oxidized metal ion, lipid peroxidation is initiated. Scavengers of O₂⁻ and H₂O₂ would affect the ratio and thereby inhibit lipid peroxidation. Similar reactions might be involved with the glucose-mediated oxidation of LDL. The marked inhibition of oxidative damage by a scavenger of lipid radicals, BHT, is consistent with inhibition of the propagation phase of the lipid peroxidation that might occur during either reaction pathway. Consistent with our observations, previous studies have demonstrated the generation of oxidants like superoxide, hydrogen peroxide, and hydroxyl radicals during nonenzymatic glycation of proteins (16, 17, 40) as well as by monosaccharide autoxidation (14–17, 40–42).

**Figure 7. Degradation of modified lipoproteins by cultured human monocyte-derived macrophages (Fig. 7a) and human skin fibroblasts (Fig. 7b).** [²⁵⁴C]labeled LDL (200 μg/ml) was modified by incubation in the absence (○), or presence of glucose (25 mM) alone (●), CuSO₄ (0.5 μM) alone (▲), or glucose and CuSO₄ together (▼) for 48 h as described in the legend to Fig. 1. After dialysis versus 150 mM NaCl, 50 μM DTPA at 4°C under N₂ to remove glucose and copper, the modified lipoproteins or acetylated LDL (□) were incubated with cultured macrophages (Fig. 7a) or skin fibroblasts (Fig. 7b) at the final concentrations indicated. After 5 h of incubation at 37°C, aliquots of medium were removed for assay of lipoprotein degradation as described in Methods. Values shown are means of triplicate determinations. The results are representative of two similar experiments.
tion of the modified lipoprotein by the LDL receptor (33, 43, 46). Degradation by fibroblasts of LDL modified in the presence of glucose or copper alone was inversely proportional to the extent of modification as assessed by TBARS and TNBS reactivity. In contrast, the uptake of the modified LDL by human monocyte-derived macrophages was enhanced with increasing oxidative modification. The extent of degradation by both fibroblasts and macrophages was related to the degree of lysine derivatization of apolipoprotein B-100. As the number of lysines reactive with TNBS decreased in macrophages, uptake and degradation increased, while fibroblasts internalized and degraded the modified lipoprotein less efficiently. These findings are consistent with uptake of the modified LDL by the macrophage scavenger receptor(s) (33) and suggest that modification of LDL induced by chronic hyperglycemia might be an important contributing factor in accelerating foam cell formation in diabetes.

Oxidized LDL has been shown to affect multiple events that might be involved in atherogenesis (11), including monocyte chemotaxis (7), cytotoxicity (47), cytokine and growth factor production (48, 49), endothelium-derived relaxing factor–mediated vascular reactivity (50), and foam cell formation (7, 9, 26, 34). If chronic hyperglycemia facilitates these events by promoting lipoprotein oxidation, this may explain in part the vascular disease observed in diabetes. Further studies are required to firmly establish whether lipoprotein oxidation is increased in patients suffering from diabetes, since these processes might be amenable to antioxidant therapy. Such antioxidants may prove to be important in preventing the markedly increased incidence of atherosclerotic complications in this common and debilitating disease.

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References


Glucose-dependent Oxidation of Low Density Lipoprotein


